

#### SECTION 1: PRODUCT AND COMPANY IDENTIFICATION

Product Name: Carbon Dioxide MSDS Manufacturer Number: 825797

Manufacturer Name: ConocoPhillips

600 N. Dairy Ashford Address:

Houston, Texas 77079-1175

www.conocophillips.com Website:

Health Issues Information: SDS@conocophillips.com

General Phone Number:

Emergency Phone Number: Chemtrec: 800-424-9300 (24 Hours)

855-244-0762

MSDS Creation Date: November 15, 2014 MSDS Revision Date: November 15, 2014

**HMIS** Health Hazard ire Hazard Reactivity 0 Personal х Protection

## SECTION 2: COMPOSITION/INFORMATION ON INGREDIENTS

**Chemical Name** CAS# **Ingredient Percent** EC Num.

Carbon dioxide 124-38-9 100% by Volume

## SECTION 3: HAZARDS IDENTIFICATION

Emergency Overview: High pressure gas.

Can cause rapid suffocation. May cause frostbite.

May cause dizziness and drowsiness

Route of Exposure: Respiratory System. Eye. Skin

Potential Health Effects: Acts as a simple asphyxiant by displacing air.

Vapor may cause a stinging sensation; liquid may cause frostbite Eye:

Skin: Liquid may cause frostbite.

Inhalation: Asphyxiant. Effects are due to lack of oxygen. Lack of oxygen can be fatal.

Moderate concentrations may cause headaches, drowsiness, dizziness, excitation, excess salivation,

vomiting, and unconciousness

Ingestion: Unlikely to occur becasue carbon dioxide is a gas at normal temperature and pressure. Liquid may cause frostbite.

Aggravation of Pre-Existing

Conditions:

None known.

# SECTION 4: FIRST AID MEASURES

Eye Contact: Immediately flush eyes with plenty of water for at least 15 to 20 minutes. Ensure adequate flushing of

the eyes by separating the eyelids with fingers. Get medical attention, if irritation or symptoms of

overexposure persists.

Skin Contact: For exposure to liquid, immediately warm frostbite area with warm water not to exceed 106 °F. In case

of an excessive exposure, remove contaminated clothing while showering with warm water. Call a

physician.

Inhalation: If inhaled, remove to fresh air. If not breathing, give artificial respiration or give oxygen by trained

personnel. Seek immediate medical attention.

Ingestion: Accidental ingestion of this material is unlikely.

#### SECTION 5: FIRE FIGHTING MEASURES

Flash Point: Not available. Flash Point Method: Not available. Auto Ignition Temperature: Not applicable. Lower Flammable/Explosive Limit: Not applicable. Upper Flammable/Explosive Limit: Not applicable.

Fire Fighting Instructions: Wear self-contained breathing apparatus and full protective clothing.

Keep fire exposed cylinders cool with water spray. If possible, stop the product flow.

Carbon Dioxide Product Code: 825797 Revision:: 11/15/2014

Extinguishing Media: Use extinguishing measures that are appropriate to local circumstances and the surrounding

Unusual Fire Hazards: Cylinder rupture may occur under fire conditions.

Hazardous Combustion

Byproducts:

Not applicable.

#### SECTION 6: ACCIDENTAL RELEASE MEASURES

Personnel Precautions: Evacuate and ventilate area

Wear a self breathing appartus (SCBA) if necessary.

Methods for containment:

Remove leaking cylinder to exhaust hood or safe outdoor area. Shut off source if possible and remove source of heat. Test for sufficient oxygen, especially in confined spaces, before allowing reentry.

Evacuate and ventilate area.

Methods for cleanup: No clean up procedures are necessary, leaks will disperse into atmosphere.

#### SECTION 7: HANDLING and STORAGE

Secure cylinder when using to protect from falling. Use suitable hand truck to move cylinders. Handling:

Use only in a well-ventilated area. Protect containers from physical damage. Do not deface cylinders or labels.

Cylinders should be refilled by qualified producers of compressed gas. Shipment of a compressed gas cylinder which has not been filled by the owner or with his written

consent is a violation of federal law (49 CFR).

Store in well ventilated areas. Storage:

Keep valve protection cap on cylinders when not in use.

Store only where temperature will not exceed 125 °F. Store full and empty cylinders separately. Use a

first-in, first-out inventory system to prevent storing full cylinders for long periods.

Refer to Compressed Gas Association (CGA) pamphlet P-1, Safe Handling of Compressed Gases in Containers, available from the CGA  $\,$ Special Handling Procedures:

Avoid eye, skin, and/or clothing contact with liquid or vapor. Emergency eyewash/shower station should Hygiene Practices:

be readily accessible. Use only in a closed system. Use piping and equipment adequately designed to withstand pressures to be encountered.

## SECTION 8: EXPOSURE CONTROLS, PERSONAL PROTECTION - EXPOSURE GUIDELINES

Provide adequate general and local exhaust ventilation to maintain concentrations below exposure **Engineering Controls:** 

Eye/Face Protection: Safety glasses

Skin Protection Description: Insulated neoprene gloves, coveralls, boots, and/or other insulated protective clothing.

Respiratory Protection: In case of leakage, use self-contained breathing apparatus.

Other Protective: Safety shoes when handling cylinders

EXPOSURE GUIDELINES

Carbon dioxide:

Guideline ACGIH: TLV-STEL: 30,000 ppm TLV-TWA: 5000 ppm Guideline OSHA: PEL-TWA: 5000 ppm

## SECTION 9: PHYSICAL and CHEMICAL PROPERTIES

Physical State Appearance: Compressed Liquefied Gas

Color: Colorless.

Sublimes -78.5 °C Boiling Point: Melting Point: Not determined.

Specific Gravity: 1 . 522 @ 0 °C (Air = 1)

Solubility: Slight

0.00198 g/ml @ 0°C Vapor Density: Vapor Pressure: 5775.2 kPa (@20°C)

Percent Volatile:

>1 (Butyl Acetate = 1) Evaporation Rate:

pH: Not determined. Molecular Weight: 44.01 g/mole Flash Point: Not available. Flash Point Method: Not available. Auto Ignition Temperature: Not applicable.

> Carbon Dioxide Revision:: 11/15/2014

Product Code: 825797

#### SECTION 10: STABILITY and REACTIVITY

Chemical Stability: Stable under normal storage conditions.

Hazardous Polymerization: Will not occur.

Conditions to Avoid: Storage in poorly ventilated areas. Storage near a heat source.

Alkali metats, alkaline earth metals, metal acetylides, chromium, titanium above 1000 °F, uranium above 1350 °F Incompatible Materials:

Special Decomposition Products: In the presence of an electrical discharge, carbon dioxide is decomposed to form carbon monoxide and

#### SECTION 11: TOXICOLOGICAL INFORMATION

#### Carbon dioxide:

 $Inhalation - Rat\ LC50 - Lethal\ concentration, 50\ percent\ kill:\ 470000\ ppm/30M\ [Details\ of\ toxic\ effects\ not\ reported\ other\ than\ lethal\ dose\ value]\ (RTECS)$ Inhalation:

#### SECTION 12: ECOLOGICAL INFORMATION

Ecotoxicity: No adverse ecological effects expected

## SECTION 13: DISPOSAL CONSIDERATIONS

Waste Disposal: Dispose of non-refillable cylinders in accordance with federal, state and local regulations. Allow gas to

vent slowly to atmosphere in an unconfined area or exhaust hood. If the cylinders are the refillable type, return cylinders to supplier with any valve outlet plugs or caps secured and valve protection caps

in place.

### SECTION 14: TRANSPORT INFORMATION

DOT Shipping Name: Compressed gases, n.o.s.

DOT UN Number: UN1013 (gas)

UN2187

DOT Hazard Class: 2.2 (nonflammable)

#### SECTION 15: REGULATORY INFORMATION

Carbon dioxide:

TSCA Inventory Status: Listed Canada DSL: Listed

## SECTION 16: ADDITIONAL INFORMATION

Label Hazard Warning: Contains gas under pressure: may explode if heated. Label Precautions: Protect from sunlight. Store in a well-ventilated place.

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MSDS Author:

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